

SOLUTION SYNTHESIS OF PROTECTED DI-2'-DEOXYNUCLEOSIDE PHOSPHOTRIESTERS VIA  
THE PHOSPHORAMIDITE APPROACH

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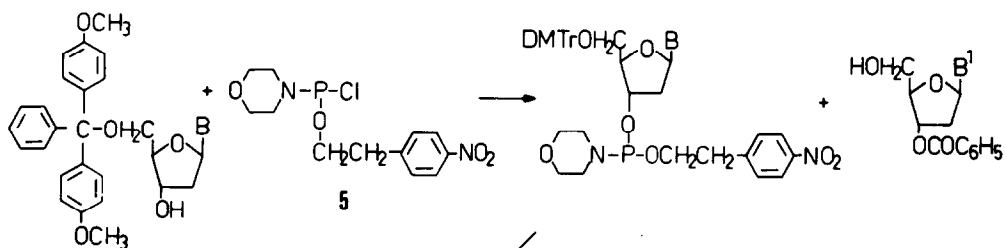
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The synthesis of fully protected di-2'-deoxyribonucleoside-3'-[2-(4-nitrophenyl)ethyl] phosphates (16-31) via the phosphoramidite approach in solution is described and the products characterized by spectrophotometrical and chromatographical means.

The newest and most successful approach in polymer supported deoxyoligonucleotide synthesis has recently been introduced by Caruthers [1,2] using deoxynucleoside phosphoramidites as a new class of highly reactive monomeric building blocks. These key intermediates are derived from appropriately protected deoxynucleosides and carry mainly the methoxy as well as the cyanoethyl group [3] as ester functions, whereas the amidite grouping varies more broadly including the dimethylamino [1,4], diethylamino [4], diisopropylamino [2,4], pyrrolidino [2], 2,2,6,6-tetramethylpiperidino [2] and morpholino [2,3,5] group.

The striking good features of the p-nitrophenylethyl (NPE) group for phosphate protection [6-8] in oligonucleotide synthesis encouraged us to investigate systematically the Caruthers method with the modified deoxynucleoside p-nitrophenylethyl phosphoromorpholidites (6-9) in solution synthesis to form with the 3'-O-benzoyl-2'-deoxyribonucleosides (10-13) the corresponding di-2'-deoxynucleoside phosphotriesters (16-31) in preparative scale.

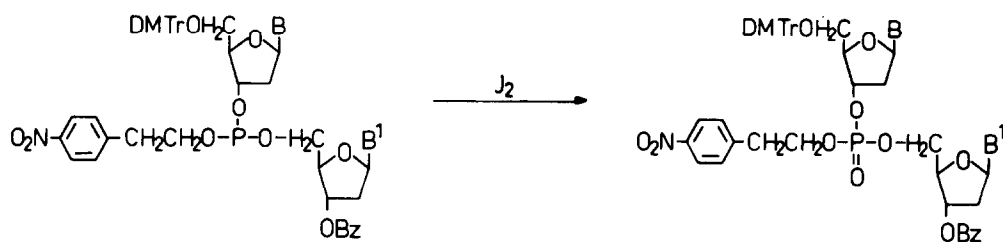
The monofunctional phosphitylating agent chloro-morpholino-2-(4-nitrophenyl) ethoxyphosphine (5) was prepared from an equimolar mixture of dichloro-2-(4-nitrophenyl)ethoxyphosphine [9] and N-trimethylsilyl-morpholino by stirring



	B
1	T
2	C
3	A
4	G

	B
6	T
7	C
8	A
9	G

	B
10	T
11	C
12	A
13	G



	B	B <sup>1</sup>
14	T	T
15	C <sup>Bz</sup>	C <sup>Bz</sup>

B	
T	
C <sup>Bz</sup>	
A <sup>Bz</sup>	
G <sup>iBu</sup>	

	B	B <sup>1</sup>
16	T	T
17	T	C <sup>Bz</sup>
18	T	A <sup>Bz</sup>
19	T	G <sup>iBu</sup>
20	C <sup>Bz</sup>	T
21	C <sup>Bz</sup>	C <sup>Bz</sup>
22	C <sup>Bz</sup>	A <sup>Bz</sup>
23	C <sup>Bz</sup>	G <sup>iBu</sup>
24	A <sup>Bz</sup>	T
25	A <sup>Bz</sup>	C <sup>Bz</sup>
26	A <sup>Bz</sup>	A <sup>Bz</sup>
27	A <sup>Bz</sup>	G <sup>iBu</sup>
28	G <sup>iBu</sup>	T
29	G <sup>iBu</sup>	C <sup>Bz</sup>
30	G <sup>iBu</sup>	A <sup>Bz</sup>
31	G <sup>iBu</sup>	G <sup>iBu</sup>

## Physical Data of Di-2'-deoxyribonucleoside Phosphotriesters

	Compound		Yield %	UV-Absorption Spectra in MeOH			HPL-Chromatography ( $R_t$ in sec.)		
	B	B <sup>1</sup>		$\lambda_{max}$ (nm)	lg $\epsilon$		CH <sub>2</sub> Cl <sub>2</sub> /MeOH (97/3) <sup>**</sup>	CH <sub>2</sub> Cl <sub>2</sub> /MeOH (80/20) <sup>**</sup>	MeOH/H <sub>2</sub> O (85/15) <sup>#</sup>
<u>16</u>	T	T	82	232	266	4.48	4.42	197/221	140
<u>17</u>	T	C <sup>Bz</sup>	79	233	262 [304]	4.66	4.61 [4.06]	161/179	183
<u>18</u>	T	A <sup>Bz</sup>	75	232	276	4.70	4.55	219/267	202
<u>19</u>	T	G <sup>iBu</sup>	81	235	261 [274]	4.68	4.54 [4.49]		399/605
<u>20</u>	C <sup>Bz</sup>	T	80	234	261 [304]	4.68	4.63 [4.10]	144/158	172
<u>21</u>	C <sup>Bz</sup>	C <sup>Bz</sup>	79	235	260 300	4.75	4.75 4.38	137/156	379
<u>22</u>	C <sup>Bz</sup>	A <sup>Bz</sup>	82	233	264	4.76	4.63	194/246	266
<u>23</u>	C <sup>Bz</sup>	G <sup>iBu</sup>	84	237	261	4.69	4.68		307/377
<u>24</u>	A <sup>Bz</sup>	T	79	233	274	4.68	4.56	197/205	167
<u>25</u>	A <sup>Bz</sup>	C <sup>Bz</sup>	81	233	263	4.77	4.66	139/147	252
<u>26</u>	A <sup>Bz</sup>	A <sup>Bz</sup>	75	233	278	4.73	4.65	185/204	298
<u>27</u>	A <sup>Bz</sup>	G <sup>iBu</sup>	76	234	261 276	4.70	4.58 4.61		297
<u>28</u>	G <sup>iBu</sup>	T	71	235	261 [274]	4.60	4.55 [4.48]		417/545
<u>29</u>	G <sup>iBu</sup>	C <sup>Bz</sup>	74	235	260	4.68	4.68		247
<u>30</u>	G <sup>iBu</sup>	A <sup>Bz</sup>	66	235	263 276	4.69	4.58 4.60		312
<u>31</u>	G <sup>iBu</sup>	G <sup>iBu</sup>	70	237	260 274	4.55	4.54 4.46		-
<u>14</u>	T	T	67	233	266	4.58	4.47		
<u>15</u>	C <sup>Bz</sup>	C <sup>Bz</sup>	64	235	260 300	4.75	4.75 4.38		

[ ] = Shoulder; \*\* ) = LiChrosorb Si 60-5, 25 cm, Chrompack;  
# ) = LiChrosorb 10 RP 8, 25 cm, Chrompack.

under N<sub>2</sub> first at -20°C and then at room temp. and followed by high vacuum evaporation to remove all volatile components. The crude, but <sup>31</sup>P-NMR spectroscopically pure material was then applied in a 1.6 molar excess to the base-protected 5'-O-dimethoxytrityl-2'-deoxyribonucleosides 1-4 to form the corresponding 2-(4-nitrophenyl) ethylphosphoromorpholidites 6-9 as colorless solids in 87-96 % isolated yields. Reaction of these intermediates towards formation

of phosphite internucleotide bonds with 3'-O-benzoyl-2'-deoxyribonucleosides (10-13) was achieved by 1H-tetrazole activation in acetonitrile followed by iodine oxidation to give the di-2'-deoxynucleoside phosphotriesters 16-31. Good yields have been obtained using a double-molar excess of the phosphoramidites 6-9 over the 5'-OH components 10-13. The intermediary 2-(4-nitrophenyl)ethyl phosphites can also be isolated and purified as shown for 14 and 15.

The characterization and structural proof of the newly synthesized compounds 16-31 was achieved by elementary analyses, UV- and <sup>1</sup>H-NMR spectrometric means as well as HPLC investigations (Tab.). It was found that there is good separation into the two diastereoisomers in the system methylenechloride/methanol (97/3) except the G<sup>iBu</sup> containing components. The guanosine moiety causes much larger retention times even in such polar systems as methylenechloride/methanol (80/20), in which 31 is still not eluted. A much better comparison of the chromatographical behaviour was, however, encountered by reversed-phase chromatography on an HPLC-column LiChrosorb 10 RP 8 in methanol/water (85/15). Separation of the diastereomeric phosphotriesters cannot be achieved under these conditions, but there is good agreement that the presence of a polar amide group in the aglycon ring effects lower retention times as expected.

#### R E F E R E N C E S

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(Received in Germany 1 February 1984)